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## Autopolycondensation of 1-Halo-2-Propanethiones. New Organic Metals and Photoconductors

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The reviously unknown 1-halo-2-propanethiones

CH<sub>3</sub>-C(C=S)-CH<sub>2</sub>X (X = F, Cl, Br) were prepared by the reaction of the corresponding propanones with hydrogen sulfide in the presence of hydrogen chloride without solvent at -70°C [1,2]. Iodo-2-propanethione (X = I) was synthe-sized by exchange reaction of 1-bromo-2-propanethione with NaI in acctone at -70°C [3].

At 20°C 1-halo-2-propanethiones (X = Cl, Br, I) undergo spontaneous polymerization to form linear polymers 1-3 with molecular mass of 1800, 1400 and 1200, respectively. Their structure is described by a general formula:

X=Cl, m=17, n=4(1); X=Br, m=14, n=0(2); X=I, m=11, n=0(3).

1-Fluoro-2-propanethione remains unchangeable even when heated to  $80^{\circ}\text{C}$  over 24 h.

The structure of polymers 1-3 was established using the methods of IR [v/cm<sup>-1</sup>: 2960, 2870 (CH<sub>3</sub>), 2730, 2690 (S-CH<sub>2</sub>), 1640 (C=C), 1220 (C=S), 750 (C-Cl), 720 (C-Br), 600 (C-I)]. UV spectroscopy [ $\lambda$ , nm: 460-500 (n $\rightarrow$  $\pi$ \* C=S), 238-300 (n $\rightarrow$  $\pi$ \*, n $\rightarrow$  $\sigma$ \* CH<sub>2</sub>=CH-S-)] and polarography. The electrochemical study of polymers 1-3 carried out on a mercury

dropping electrode in CH<sub>3</sub>CN against 0.05 M Bu<sub>4</sub>NClO<sub>4</sub> background, ha revealed the presence of (C=S) group (0.017v, 8.8 μA×ml/mg).

As shown by ESR spectroscopy, polymers 1-3 are paramagnetic [g-2.0062, H = 0.82 mT,  $N = 0.93 \times 10^{17} \text{ spin/g. 1}$ ; 2.0069, 0.80,  $1.00 \times 10^{-17} \text{ 2}$  2.0060, 1.42,  $1.60 \times 10^{-17} \text{ 3}$ ]. The signal parameters are typical of poly conjugated systems.

The following scheme describing the formation of polymers 1-3 i suggested:

$$S=C-CH_{2}X \Longrightarrow HS-C=CHX \xrightarrow{S=C(CH_{1})-CH_{2}X} S=C-CH_{2}-S-C=CHX \xrightarrow{-HX} CH_{3} CH_{3} CH_{3}$$

$$\xrightarrow{k \text{ HS-C(CH_{3})-CHX}} S=C-CH_{2}-[-S-C=CH-]_{k}-S-C=CHX \xrightarrow{(a+1) \text{ HX}} CH_{3} CH_{3} CH_{3}$$

$$\longrightarrow S=C-CH_{2}-[-S-C=CH-]_{m}-[-S-CX-CH_{2}-]_{n}-S-CH-CHX_{2}$$

$$CH_{3} CH_{3} CH_{3} CH_{3} CH_{3}$$

The electroconduction (in air and vacuum,  $\sigma$  Cm/cm, 1:  $2.1\times10^{-14}$   $1.0\times10^{-14}$ ; 2:  $6.5\times10^{-14}$ ,  $4.1\times10^{-14}$ ; 3:  $2.6\times10^{-8}$ ,  $1.2\times10^{-8}$ ), photoconduction ( $I_{photo}$  /  $I_{dark}$ , relative units,  $\lambda_{max}$ , nm, 1: 12, 460; 2: 11, 450) o polymers 1-3 and ESR data provide evidence for the presence of developed conjugation system and allow assignment of these polymers to organic semiconductors 1-2 and to organic metals 3.

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